



**Advance  
Chemicals Ltd.**

# Safety Data Sheet

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## Section 01 - Product And Company Identification

<b>Product Identifier</b>	Advance 12A
<b>UN/ID No</b>	UN 1791
<b>Synonyms</b>	12% Chlorine Bleach, Sodium Hypochlorite Solutions, Javel Water, Liquid Swimming Pool
<b>Formula</b>	NaOCl
<b>Molecular Weight</b>	74.44
<b>Product Use</b>	Disinfectant, sanitizer, odour control, water purification, textile bleaching, commercial laundry applications.
<b>Supplier Name</b>	Advance Chemicals Ltd A Division of ClearTech Industries Inc. 1500 Quebec Avenue Saskatoon, SK. Canada S7K 1V7
<b>Prepared By</b>	ClearTech Industries Inc. Technical Department Phone: 1(800)387-7503
<b>24-Hour Emergency Phone</b>	1(800)387-7503

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## Section 02 - Hazard Identification

### GHS-Classification

<b>Skin Corrosion/Irritation</b>	Category 1B
<b>Serious Eye Damage/Eye Irritation</b>	Category 1
<b>Health Hazards not elsewhere classified (corrosive)</b>	Category 1

**Signal Word**

Danger

**Hazard Statements**

Causes severe skin burns and eye damage

Causes serious eye damage

**Pictograms****Precautionary Statements**

Immediately call a POISON CENTER or doctor/physician.

Do not breathe dusts or mists

Wash hands and exposed areas thoroughly after handling

Wear protective gloves and eye/face protection

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower.

Wash contaminated clothing before reuse.

IF INHALED: Remove to fresh air and keep at rest in a position comfortable for breathing.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Store locked up

Dispose of contents/container in accordance with Federal, Provincial, Municipal or other applicable regulations

For specific treatment refer to Section 4 of SDS

**Section 03 - Composition / Information on Ingredients**

Component	CAS Number	Weight %	Remarks
Sodium Hypochlorite	7681-52-9	10-12%	

**Section 04 - First Aid Measures**

<b>Inhalation</b>	Remove victim to fresh air. Give artificial respiration only if breathing has stopped. If breathing is difficult, give oxygen. Seek immediate medical attention.
<b>Skin Contact / Absorption</b>	As quickly as possible, flush with lukewarm, gently flowing water for at least 20 minutes, or until the chemical is removed. If irritation persists, repeat flushing. Under running water, remove contaminated clothing, shoes and leather goods. Completely decontaminate clothing, shoes and leather goods before reuse, or discard. Obtain medical advice immediately.
<b>Eye Contact</b>	Check for and remove any contact lenses. Flush immediately with water for at least 20 minutes. Forcibly hold eyelids apart to ensure complete irrigation of eye tissue. Seek immediate medical attention.
<b>Ingestion</b>	NEVER give anything by mouth if victim is rapidly losing consciousness, is unconscious or convulsing. Have victim rinse mouth thoroughly with water. DO NOT INDUCE VOMITING. Have victim drink 240 to 300 mL (8 to 10 oz) of water to dilute material in stomach. If vomiting occurs naturally, have victim lean forward to reduce risk of aspiration. Rinse mouth and repeat administration of water. Quickly transport victim to an emergency care facility.
<b>Additional Information</b>	Provide general supportive measures (comfort, warmth, rest). Consult a doctor and/or the nearest Poison Control Centre for all exposures except minor instances of inhalation or skin contact.

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**Section 05 - Fire Fighting Measures**


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<b>Flammable Properties</b>	Non-Flammable
<b>Flash Point</b>	Not Applicable
<b>Auto-ignition Temperature</b>	Not Applicable
<b>Upper Flammable Limit</b>	Not Applicable
<b>Lower Flammable Limit</b>	Not Applicable
<b>Explosive Properties</b>	Pressure buildup in containers could result in an explosion when heated or in contact with acidic fumes. Vigorous reaction with oxidizable organic materials may result in a fire.
<b>Suitable Extinguishing Media</b>	Product does not burn. Use appropriate extinguishing media for material that is supplying the fuel to the fire.
<b>Specific Hazards During Fire Fighting</b>	Chlorine, hydrogen chloride gas, oxygen gas and disodium oxide.
<b>Special Protective Equipment for Fire-Fighters</b>	Wear NIOSH-approved self-contained breathing apparatus and protective clothing.
<b>Further Information</b>	DO NOT use dry chemical fire extinguishing agents containing ammonium compounds (such as some A:B:C agents), since an explosive compound can be formed.

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**Section 06 - Accidental Release Measures**


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<b>Personal Precautions</b>	Wear appropriate personal protective equipment. Ventilate area. Only enter area with PPE. Stop or reduce leak if safe to do so. Prevent material from entering sewers. Flush with water to remove any residue.
<b>Environmental Precautions</b>	Prevent material from entering sewers or confined spaces.
<b>Methods For Cleaning Up</b>	<p>Contain spill with earth, sand or absorbent material which does not react with spilled material.</p> <p>SMALL SPILLS should be wiped up with absorbent material and disposed of in government approved waste containers. Paper towels may become extremely hot when saturated with this product and cause a secondary fire hazard. Rinse out absorbent materials used in small spill recovery with plenty of water.</p> <p>LARGER SPILLS should be contained by diking with sand, soil or other absorbent, non-combustible material, then transferred into approved waste containers for proper disposal. Keep product out of sewers, storm drains, surface run-off water and soil. Harmful to aquatic life at low concentrations. Can be dangerous if allowed to enter potable water intakes. Wear appropriate respiratory protection and restrict access to non-protected personnel. Comply with all government regulations on spill reporting, and handling and disposal of waste.</p> <p>The contained bleach spill can be effectively neutralized as follows;</p> <ol style="list-style-type: none"> <li>1. Wear respiratory protection and protective clothing, gloves, glasses, etc.</li> <li>2. Very slowly and cautiously, apply a dilute aqueous solution of Sodium Sulphite, or Sodium meta-Bisulphite to the spill. Mix well. This neutralizes the available chlorine content while reducing the pH to about pH 4. Check with a pH meter or test strip paper. Chlorine gas is a dangerous by-product of this reaction procedure.</li> <li>3. Increase the pH of the contained spill to about pH 7 by slowly adding a dilute aqueous solution of Soda Ash or Sodium Bicarbonate. Check pH frequently.</li> <li>4. The bleach spill should be neutral, with a pH of 7. Check with the appropriate local, provincial or federal agencies for proper and correct disposal methods for this product. If available, use a field test kit to check for levels of residual chlorine in the treated waste water.</li> </ol>

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**Section 07 - Handling and Storage**


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<b>Handling Procedures</b>	Use proper equipment for lifting and transporting all containers. Use sensible industrial hygiene and housekeeping practices. Wash thoroughly after handling. Avoid all situations that could lead to harmful exposure. Avoid generating mists. Prevent the release of mists into the workplace air. Never add water to a corrosive. Always add corrosives to water. When mixing with water, stir small amounts in slowly. Use cold water to prevent excessive heat generation. Never return contaminated material to its original container.
<b>Storage Requirements</b>	Store in a cool, dry, well-ventilated area, out of direct sunlight and away from heat sources. Strong solutions (greater than 10% available chlorine) may slowly give off chlorine during storage, especially when warm (above 18°C). Vent caps may be required to prevent a build-up of pressure that could cause containers to burst.
<b>Incompatible Materials</b>	Primary amines, aromatic amines, ammonium salts, ammonia, urea, phenylacetonitrile, acids, metals, reducing agents, ethyleneimine, methanol, formic acid, furfuraldehyde, ethanediol, sodium ethylenediaminetetracetate solution and sodiumhydroxide solution.

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**Section 08 - Exposure Controls and Personal Protection**


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**Exposure Limit(s)**

Component	Regulation	Type of Listing	Value
Sodium hypochlorite	AIHA	WEEL-STEL	2mg/m <sup>3</sup> (15 min)
Chlorine	ACGIH	TLV-TWA	0.5ppm

**Engineering Control(s)**

**Ventilation Requirements** Mechanical ventilation (dilution or local exhaust), process or personnel enclosure and control of process conditions must be provided in accordance with all fire codes and regulatory requirements. Supply sufficient replacement air to make up for air removed by exhaust systems.

**Other** Emergency shower and eyewash must be available and tested in accordance with regulations and be in close proximity.

**Protective Equipment**

**Eyes/Face** Chemical goggles, full-face shield, or a full-face respirator is to be worn at all times when product is handled. Contact lenses should not be worn; they may contribute to severe eye injury.

**Hand Protection** Impervious gloves of chemically resistant material (rubber or PVC) should be worn at all times. Wash contaminated clothing and dry thoroughly before reuse.

**Skin and Body Protection** Body suite, aprons, and/or coveralls of chemical resistant material should be worn at all times. Wash contaminated clothing and dry thoroughly before reuse. Impervious boots of chemically resistant material should be worn at all times. No special footwear is required other than what is mandated at place of work.

**Respiratory Protection** A NIOSH-approved respirator suitable for chlorine is recommended. Where a higher level of protection is required, use a self-contained breathing apparatus.

**Thermal Hazards** Not Available

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**Section 09 - Physical and Chemical Properties**


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**Appearance**

<b>Physical State</b>	Liquid
<b>Colour</b>	Clear, greenish-yellow solution.
<b>Odour</b>	Strong chlorine odour.
<b>Odour Threshold</b>	Not Available

**Property**

<b>pH</b>	< 12
<b>Melting Point/Freezing Point</b>	~ -15°C (12% trade)
<b>Initial Boiling Point and Boiling Range</b>	Decomposes
<b>Evaporation Rate</b>	Not Available
<b>Vapour Pressure (mm Hg, 20°C)</b>	12.1 mmHg at 20°C (12.5 wt. %)
<b>Vapour Density (Air=1)</b>	Not Available
<b>Relative Density</b>	Not Available
<b>Solubility(ies)</b>	Completely soluble in water.
<b>Partition Coefficient: n-octanol/water</b>	Log P <sub>OW</sub> = -3.42 (estimated)
<b>Auto-ignition Temperature</b>	Not Applicable
<b>Decomposition Temperature</b>	Slowly decomposes above 40°C
<b>Viscosity</b>	Not Available
<b>Specific Gravity (Water=1)</b>	1.16-1.17
<b>% Volatiles by Volume</b>	Not Available

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**Section 10 - Stability and Reactivity**


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<b>Reactivity</b>	Sodium hypochlorite solutions decompose slowly at normal temperatures releasing low concentrations of corrosive chlorine gas. Decomposition is influenced by temperature, concentration, pH, ionic strength, exposure to light and the presence of metals, such as copper, nickel or cobalt, metal oxides, e.g. rust and other impurities, such as acids and amines. Hypochlorites react with urea to form nitrogen trichloride which explodes spontaneously in air.
<b>Stability</b>	Unstable at temperatures above 40°C, in sunlight, and in contact with acid.
<b>Incompatible Materials</b>	Primary amines, aromatic amines, ammonium salts, ammonia, urea, phenylacetonitrile, acids, metals, reducing agents, ethyleneimine, methanol, formic acid, furfuraldehyde, ethanediol, sodium ethylenediaminetetracetate solution and sodiumhydroxide solution.
<b>Conditions to Avoid</b>	Heat, sunlight, acidic conditions, the presence of metals and other impurities.
<b>Hazardous Decomposition Products</b>	Chlorine (by reaction with acids), oxygen (by reaction with nickel, copper, tin, manganese, iron), sodium chloride, sodium chlorate, with increased temperature.
<b>Possibility of Hazardous Reactions</b>	Hazardous polymerization will not occur. Reacts exothermically with acids. Reacts with ammonia, amines and ammonia salts to produce chloramines. Decomposes on heating to produce chlorine gas.

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**Section 11 - Toxicological Information**


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**Acute Toxicity**

Component	Oral LD <sub>50</sub>	Dermal LD <sub>50</sub>	LC <sub>50</sub>
Undiluted Sodium Hypochlorite	8910mg/kg (rat) 5800mg/kg (mouse)	> 10,000mg/kg (rabbit)	5250mg/m <sup>3</sup> (rat, 4hr exposure)

**Chronic Toxicity – Carcinogenicity**

Component	IARC
Sodium Hypochlorite	Group 3: Not classifiable as to its carcinogenicity to humans.

**Skin Corrosion/Irritation**

Very dilute solutions have caused negligible irritation, while more concentrated solutions have caused corrosive injury.

**Serious Eye Damage/Irritation**

Very dilute solutions have caused no irritation. More concentrated solutions have caused corrosive injury, which did not heal within 21 days.

**Respiratory or Skin Sensitization**

Not known to be a respiratory or skin sensitizer.

**Germ Cell Mutagenicity**

The available information does not suggest that sodium hypochlorite is mutagenic.

**Reproductive Toxicity**

Not Available

**STOT-Single Exposure**

May cause respiratory irritation.

**STOT-Repeated Exposure**

Not Available

**Aspiration Hazard**

Prolonged or repeated overexposure causes lung damage.

**Synergistic Materials**

Not Available

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**Section 12 - Ecological Information**


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**Ecotoxicity**

Component	Toxicity to Algae	Toxicity to Fish	Toxicity to Daphnia and Other Aquatic Invertebrates
Sodium Hypochlorite	EC <sub>50</sub> (Dunaliella sp., 24hr): 0.4mg/L EC <sub>50</sub> (Dunaliella tertiolecta, 24hr): 0.11mg/L	LC <sub>50</sub> (Clupea harengus, 96hr): 0.033-0.097mg/L LC <sub>50</sub> (Oncorhynchus gorbuscha, 96hr): 0.023- 0.052mg/L	EC <sub>50</sub> (Daphnia magna, 96hr): 2.1mg/L LC <sub>50</sub> (Gammarus fasciatus, 96hr): 4mg/L
<b>Biodegradability</b>	Not Available		
<b>Bioaccumulation</b>	Not Available		
<b>Mobility</b>	Not Available		

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**Section 13 - Disposal Considerations**


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<b>Product, Wastes and Packaging</b>	Dispose of product and containers in accordance with all federal, provincial and municipal regulations.
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**Section 14 - Transport Information**


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<b>UN Number</b>	UN 1791
<b>UN Proper Shipping Name</b>	Hypochlorite Solution more than 7% available Chlorine
<b>Transport Hazard Class(es)</b>	8
<b>Packaging Group</b>	III
<b>Environmental Hazards</b>	Not listed as a marine pollutant under Canadian TDG Regulations Schedule 3, Column 4.
<b>Special Precautions</b>	
<b>Transport in Bulk</b>	
<b>TDG</b>	
<b>Other</b>	Secure containers (full and/or empty) with suitable hold down devices during shipment and ensure all caps, valves, or closures are secured in the closed position.

**PRODUCT CLASSIFICATION:** This product has been classified on the preparation date specified at section 16 of this MSDS / SDS, for transportation in accordance with the requirements of part 2 of the Transportation of Dangerous Goods Regulations. If applicable, testing and/or published test data regarding the classification of this product are listed in the references at section 16 of this MSDS / SDS.

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**Section 15 - Regulatory Information**


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<b>NSF Certification</b> .....	Product is certified under NSF/ANSI Standard 60 for disinfection and oxidation at a maximum dosage for the following:
	sodium hypochlorite 5%: 174mg/L
	sodium hypochlorite 6%: 145mg/L
	sodium hypochlorite 7%: 125mg/L
	sodium hypochlorite 8%: 109mg/L
	sodium hypochlorite 9%: 97mg/L
	sodium hypochlorite 10%: 87mg/L
	sodium hypochlorite 11%: 79mg/L
	sodium hypochlorite 12%: 72mg/L
	sodium hypochlorite 13%: 67mg/L
	sodium hypochlorite 14%: 62mg/L
	sodium hypochlorite 15%: 58mg/L
	sodium hypochlorite 16%: 55mg/L
	sodium hypochlorite 17%: 51mg/L
	sodium hypochlorite 18%: 48mg/L
	sodium hypochlorite 19%: 46mg/L
	sodium hypochlorite 20%: 43mg/L

**NOTE: Any product strength below 7% is not regulated by TDG.**

**Sanitizer Use:** to obtain 10 liters of a 200 mg/L solution as available chlorine, use 16.7 mL of Hypochlor-12 for each 10 liters of clean, potable water.

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**Section 16 - Other Information**


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**Preparation Date** August 11, 2014

**Note:** The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations.

**Attention: Receiver of the chemical goods / MSDS coordinator**

As part of our commitment to the Canadian Association of Chemical Distributors (CACD) Responsible Distribution<sup>®</sup> initiative, ClearTech Industries Inc. and its associated companies require, as a condition of sale, that you forward the attached Material Safety Data Sheet(s) to all affected employees, customers, and end-users. ClearTech will send any available supplementary handling, health, and safety information to you at your request.

If you have any questions or concerns please call our customer service at 1(800)387-7503.

The Product is certified Kosher

**Abbreviations**

ACGIH	American Conference of Governmental Industrial Hygienists
AIHA	American Industrial Hygiene Association
CAS	Chemical Abstract Service
DSL	Domestic Substance List
EC	Effective Concentration
IARC	International Agency for Research on Cancer
IDHL	International Digest of Health Legislation
LC	Lethal Concentration
LD	Lethal Dosage
MSHA	Mine Safety and Health Administration
NIOSH	National Institute for Occupational Safety and Health
NFPA	National Fire Protection Association
NTP	National Toxicology Program (U.S.A.)
OSHA	Occupational Safety and Health Administration (U.S.A.)
PEL	Permissible Exposure Limit
PPE	Personal Protection Equipment
STEL	Short-term Exposure Limit
STOT	Specific Target Organ Systemic Toxicity
TLV	Threshold Limit Value
TSCA	Toxic Substances Control Act
TWA	Time Weighted Average
UN	United Nations
WEEL	Workplace Environmental Exposure Level
WHMIS	Workplace Hazardous Materials Information System

**References**

- 1) CHEMINFO: Sodium hypochlorite solutions. (2014). Retrieved from Canadian Centre for Occupational Health and Safety: <http://ccinfoweb2.ccohs.ca/cheminfo/records/351E.html>
- 2) Safety Data Sheet. (2013, January 22). Retrieved from Orica Chemicals: <http://msds.orca.com/pdf/shess-en-cds-010-000034421401.pdf>
- 3) U.S. National Library of Medicine. (2003). Sodium hypochlorite. Retrieved from TOXNET: <http://toxnet.nlm.nih.gov/cgi-bin/sis/search/f?./temp/~1M5jou:1>
- 4) <http://echa.europa.eu/information-on-chemicals/cl-inventory-database/-/cl-inventory/view-notification-details/152374/37061036>
- 5) [http://www.csst.qc.ca/prevention/reptox/pages/fiche-complete.aspx?no\\_produit=41517&nom\\_produit=Hypochlorite%20de%20sodium%2012%](http://www.csst.qc.ca/prevention/reptox/pages/fiche-complete.aspx?no_produit=41517&nom_produit=Hypochlorite%20de%20sodium%2012%)

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**ClearTech Industries Inc. - Locations**


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[www.ClearTech.ca](http://www.ClearTech.ca)

<b>Location</b>	<b>Address</b>	<b>Postal Code</b>	<b>Phone Number</b>
Port Coquitlam, B.C.	223 Kingsway Avenue	V3C 1S9	1(800)387-7503
Calgary, AB.	5516E - 40 <sup>th</sup> St. S.E.	T2C 2A1	1(800)387-7503
Edmonton, AB.	12020 - 142 <sup>nd</sup> Street	T5L 2G8	1(800)387-7503
Saskatoon, SK.	North Corman Industrial Park	S7K 1V7	1(800)387-7503
Regina, SK.	555 Henderson Drive	S42 5X2	1(800)387-7503
Winnipeg, MB.	340 Saulteaux Crescent	R3J 3T2	1(800)387-7503
Mississauga, ON.	7480 Bath Road	L4T 1L2	1(800)387-7503

**24 Hour Emergency Number - All Locations – 1(800)387-7503**

**End of Safety Data Sheet**